

ATOMS

Q.1 Complete the table

Particle	Relative charge	Relative mass	Position
PROTON			
NEUTRON			
ELECTRON			

9

Q.2 Define Atomic Number

1

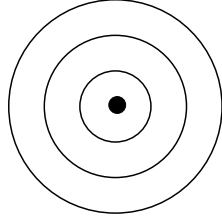
Q.3 Define Mass Number

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Q.4 Write out the electronic configurations of the following species

Li K H
 Al S Ar 6

Q.5 Using crosses for electrons, draw out the electronic configuration of chlorine atoms (atomic no 17).



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Q.6 What do the electronic configurations of He, Ne and Ar have in common?

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Q.7 What do the electronic configurations of Li, Na and K have in common?

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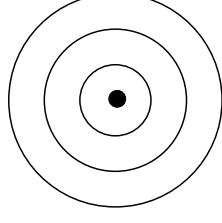
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